(FILE 'HOME' ENTERED AT 10:38:49 ON 20 APR 2007). FILE 'CAPLUS' ENTERED AT 10:38:57 ON 20 APR 2007 FILE 'REGISTRY' ENTERED AT 10:39:08 ON 20 APR 2007 L1 1 S ACETALDEHYDE/CN L2 1 S METHYL IODIDE/CN FILE 'CAPLUS' ENTERED AT 10:40:37 ON 20 APR 2007 S 75-07-0/REG# AND 74-88-4/REG# FILE 'REGISTRY' ENTERED AT 10:41:18 ON 20 APR 2007 L3 1 S 74-88-4/RN FILE 'CAPLUS' ENTERED AT 10:41:18 ON 20 APR 2007 L420289 S L3 FILE 'REGISTRY' ENTERED AT 10:41:19 ON 20 APR 2007 L5 1 S 75-07-0/RN FILE 'CAPLUS' ENTERED AT 10:41:19 ON 20 APR 2007 L6 35838 S L5 536 S L6 AND L4 L7 L8 9 S L7 AND (DENSITY OR HEATING RATE OR COLUMN PRESSURE OR FEED C L9 4 S L8 AND PY<2004 FILE 'STNGUIDE' ENTERED AT 10:44:56 ON 20 APR 2007 FILE 'CAPLUS' ENTERED AT 10:48:02 ON 20 APR 2007 L10 5250 S 75-07-0/PREP L11 33 S 75-07-0/PUR L12 3497 S 75-07-0/PROC L13 8637 S L10 OR L11 OR L12 L14 224 S L13 AND DISTIL? L15 12 S L14 AND (DENSITY OR HEATING RATE OR COLUMN PRESSURE OR FEED

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L16

11 S L15 AND PY<2004

=> d 1-11 ibib abs hitstr

L16 ANSWER 1 OF 11 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1997:218538 CAPLUS

DOCUMENT NUMBER: 126:211821

TITLE: Method for separation and removal of acetaldehyde by

distillation in manufacture of acetic acid

INVENTOR(S): Morimoto, Yoshiaki; Nakayama, Hiroyuki

PATENT ASSIGNEE(S): Daicel Chem, Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 7 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent. LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 09040590	A	19970210	JP 1995-198290	19950803 <
JP 3883221	B2	20070221		
JP 2006096764	Α	20060413	JP 2005-335002	20051118
PRIORITY APPLN. INFO.:			JP 1995-198290	A3 19950803
AB Acetaldehyde is sep	arated	and removed	from a mixture liquid	
aceralgenyde				

AB acetaldehyde

 \leq 5, Me acetate \leq 10, acetic acid 20-50, MeI 0.1-20, and H20

5-50 weight% by distillation using a distillation tower having ≥40 plates at reflux ratio ≥ 10 . At least either one of the

phase containing Me iodide as the main component and the phase containing H2O as

the main component is refluxed to said distillation tower while maintaining the liquid fractionation state of the liquid distilled at the distillation tower. This

process efficiently separates acetaldehyde and MeI from a mixture liquid containing acetaldehyde, MeI, Me acetate, and acetic acid and is suitable for purification of acetic acid obtained by carbonylation of methanol.

75-07-0, Acetaldehyde, processes

RL: REM (Removal or disposal); PROC (Process)

(method for separation and removal of acetaldehyde by distillation in manufacture of

acetic acid)

RN 75-07-0 CAPLUS

Acetaldehyde (CA INDEX NAME)

 $H_3C-CH=0$

L16 ANSWER 2 OF 11 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1987:86875 CAPLUS

DOCUMENT NUMBER: 106:86875

TITLE: A simple exact method for calculating tangent pinch

points in multicomponent nonideal mixtures by

bifurcation theory

AUTHOR(S): Levy, Sanford G.; Doherty, Michael F.

CORPORATE SOURCE: Chem. Eng. Dep., Univ. Massachusetts, Amherst, MA,

01003, USA

SOURCE: Chemical Engineering Science (1986), 41(12),

3155-60

CODEN: CESCAC; ISSN: 0009-2509

DOCUMENT TYPE: Journal LANGUAGE: English

The tangent pinch problem for multicomponent systems was studied.

notion of a binary tangent pinch logically extends to the multicomponent case. A bifurcation anal. provides further insight into the math. conditions that apply at a tangent pinch, along with a procedure which serves to locate the min. reflux ratio under tangent pinch conditions.

IT 75-07-0P, Acetaldehyde, preparation

RL: PREP (Preparation)

(distillation of, tangent pinch point in, calcn. of, by bifurcation theory)

RN 75-07-0 CAPLUS

CN Acetaldehyde (CA INDEX NAME)

 $H_3C-CH=0$

L16 ANSWER 3 OF 11 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1976:476278 CAPLUS

DOCUMENT NUMBER: 85:76278

TITLE: A treatment of grape wine distillation heads
AUTHOR(S): Williams, Patrick J.: Strauss, Christopher P.

AUTHOR(S): Williams, Patrick J.; Strauss, Christopher R.
CORPORATE SOURCE: Aust. Wine Res. Inst., Glen Osmond, Australia
SOURCE: Journal of the Science of Food and Agriculture (

1976), 27(6), 487-98

CODEN: JSFAAE; ISSN: 0022-5142

DOCUMENT TYPE: Journal LANGUAGE: English

AB SO2 [7446-09-5], MeCHO [75-07-0], acetals, and MeCO2Et [141-78-6] were eliminated from grape wine distillation heads by inert gas sparging of the diluted

and acidified heads held at reflux until an acceptable concentration of MeCHO was reached. The treated product, obtained after basification and distillation was as good as high-quality fortifying spirit. The EtOH [64-17-5] recovery of this process, designed for use in a modified pot still, is greater than that obtained by a comparable fractional distillation of heads.

IT 75-07-0, uses and miscellaneous

RL: REM (Removal or disposal); PROC (Process) (removal of, from grape wine distillation heads)

RN 75-07-0 CAPLUS

CN Acetaldehyde (CA INDEX NAME)

H3C-CH-0

L16 ANSWER 4 OF 11 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1966:482140 CAPLUS

DOCUMENT NUMBER: 65:82140
ORIGINAL REFERENCE NO.: 65:15322d-f

TITLE: Removal of acetaldehyde from ethylene oxide by

extractive distillation

INVENTOR(S): Leis, Donald G.; Mills, Edward J., Jr.

PATENT ASSIGNEE(S): Union Carbide Corp.

SOURCE: 3 pp.
DOCUMENT TYPE: Patent

LANGUAGE: Facence Unavailable

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 3265593		19660809	US 1962-229160	19621008 <
PRIORITY APPLN. INFO.:			US	19621008

=> d 1-11 ibib abs hitstr

L16 ANSWER 1 OF 11 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1997:218538 CAPLUS

DOCUMENT NUMBER: 126:211821

TITLE: Method for separation and removal of acetaldehyde by

distillation in manufacture of acetic acid

INVENTOR(S): Morimoto, Yoshiaki; Nakayama, Hiroyuki

PATENT ASSIGNEE(S): Daicel Chem, Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 7 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 09040590	A	19970210	JP 1995-198290	19950803 <-
JP 3883221	B2	20070221	•	
JP 2006096764	Α	20060413	JP 2005-335002	20051118
PRIORITY APPLN. INFO.:			JP 1995-198290	A3 19950803
AB Acetaldehyde is se	eparated	and removed	from a mixture liqu	id containing
acetaldehyde			-	3

 \leq 5, Me acetate \leq 10, acetic acid 20-50, MeI 0.1-20, and H20

5-50 weight% by distillation using a distillation tower having \geq 40 plates at reflux ratio \geq 10. At least either one of the

phase containing Me iodide as the main component and the phase containing H2O as

the main component is refluxed to said distillation tower while maintaining the liquid fractionation state of the liquid distilled at the distillation tower. This

process efficiently separates acetaldehyde and MeI from a mixture liquid containing acetaldehyde, MeI, Me acetate, and acetic acid and is suitable for purification of acetic acid obtained by carbonylation of methanol.

IT 75-07-0, Acetaldehyde, processes

RL: REM (Removal or disposal); PROC (Process)

(method for separation and removal of acetaldehyde by distillation in manufacture of

acetic acid)

RN 75-07-0 CAPLUS

CN Acetaldehyde (CA INDEX NAME)

 $H_3C-CH=0$

L16 ANSWER 2 OF 11 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1987:86875 CAPLUS

DOCUMENT NUMBER: 106:86875

TITLE: A simple exact method for calculating tangent pinch

points in multicomponent nonideal mixtures by

bifurcation theory

AUTHOR(S): Levy, Sanford G.; Doherty, Michael F.

CORPORATE SOURCE: Chem. Eng. Dep., Univ. Massachusetts, Amherst, MA,

01003, USA

SOURCE: Chemical Engineering Science (1986), 41(12),

3155-60

CODEN: CESCAC; ISSN: 0009-2509

DOCUMENT TYPE: Journal

LANGUAGE: English

AB The tangent pinch problem for multicomponent systems was studied. The

Low mol. weight alcs., diols, ethers, ether-alcs., polyalkylene glycols, AB esters, alkyl isobutyl ketones, and alkanoxy alcs. which boil at 65-120°, which are miscible with both ethylene oxide (I) and AcH, and which are inert to I serve for the extraction of AcHO impurities from I under distillation conditions. In an example, using a distillation apparatus consisting of

a kettle equipped with an Oldershaw column having 3 sections of 15 trays each, an overhead condenser, and a receiver, 580 ml./hr. of I containing 435 ppm. AcH was fed in at the 15th tray from the kettle as 116 ml./hr. MeOH $\,$ was introduced at the 30th tray from the kettle. At a reflux ratio of 0.33:1, 10.5° overhead temperature, and atmospheric pressure, 89.4% of the I added was collected overhead as MeOH containing 92% of the AcH contaminant was removed from the kettle. When the I-MeOH ratio was increased to 8:1, the AcH content of the product was reduced by only 78%. Other extractants used were EtOH, Me Cellosolve, Bu Cellosolve, Me iso-Bu ketone, 2-ethylhexyl acetate, diiso-Bu ketone, and Cellosolve acetate.

TT 75-07-0P, Acetaldehyde RL: PREP (Preparation)

(separation of, from ethylene oxide)

RN 75-07-0 CAPLUS

CNAcetaldehyde (CA INDEX NAME)

 $H_3C-CH=0$

L16 ANSWER 5 OF 11 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1964:66319 CAPLUS

DOCUMENT NUMBER: 60:66319 ORIGINAL REFERENCE NO.: 60:11639e-q

TITLE:

Method for a continuous separation of concentrated

acetic acid from an acetaldehyde oxidation reaction

INVENTOR(S): Subr, Milan; Coufal, Jiri

SOURCE: 7 pp. DOCUMENT TYPE: Patent LANGUAGE: Unavailable

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

KIND DATE PATENT NO. APPLICATION NO. --------------------CS 102869 19620315 CS 19601230 <--PRIORITY APPLN. INFO.: CS

The economy and the quality of the com. process are improved (1) by utilizing the heat of condensation of the separated concentrated HOAc to heat the

boiler of one of the refluxing columns, and (2) by returning the unreacted ACH for further oxidation, and the dilute HOAc yields for further refluxing. In this method, the distillate emanating from the head of the 1st refluxing column, which is initially charged with a preheated AcH oxidation reaction mixture, is led to the 2nd refluxing column, and as a side product, vaporized concentrated HOAc is removed at a lower level of the 1st column and piped to the boiler of the 2nd refluxing column. There, it condenses, releasing heat which heats the column. The distillate from the head of the 2nd refluxing column enters the 3rd column, where it is separated into 3 fractions; the lightest, which is returned to the AcH oxidation station, the middle, which is further processed, and the HOAc fraction which is combined with the initial charge to the 1st column. The boiler residues from the 2nd column are continuously returned to the 1st column; those from the 1st column are returned, in the liquid state, to the AcH oxidation station. The concentrated (99.85% by weight) HOAc end

after it gives up its heat of condensation to heat the 2nd refluxing column, is further cooled and pumped into storage. The utilization of this heat of condensation enables the 2nd column to operate at a high reflux ratio, as a result of which the boiler residues from the 2nd column attain the same (or higher) HOAc concentration as that possessed by the raw acid and can be fed directly back into the 1st column. Thus, the formation of undesirable dilute HOAc is avoided and the content of harmful ethylidene diacetate is held to a min.

IT 75-07-0P, Acetaldehyde

RL: PREP (Preparation)

(oxidation of, distillation of product of)

RN 75-07-0 CAPLUS

CN Acetaldehyde (CA INDEX NAME)

 $H_3C-CH=0$

L16 ANSWER 6 OF 11 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1961:136172 CAPLUS

DOCUMENT NUMBER: 55:136172

ORIGINAL REFERENCE NO.: 55:25600i,25601a-c

DETERMINE REFERENCE NO.: 55:250001,25001a-C

TITLE: Microdetermination of acetals of acetaldehyde, vinyl

ethers, and other compounds containing combined

acetaldehyde groups

AUTHOR(S): Bowman, Malcolm C.; Beroza, Morton; Acree, Fred, Jr.

CORPORATE SOURCE: U.S. Dept. of Agr., Orlando, FL SOURCE: Anal. Chem. (1961), 33, 1053-5

CODEN: ANCHAM; ISSN: 0003-2700

DOCUMENT TYPE: Journal LANGUAGE: Unavailable

AB A microdetn. of Sesamex or other compds. that yield AcH upon acid hydrolysis was developed. The hydrolysis apparatus was the all-glass assembly described by Giang and Smith (CA 50, 16027c), which permits reflux and subsequent distillation without interrupting to transfer material. CHCl3 solns. of samples were obtained by diluting 200 $\mu l.$ of each to 50 ml. in volumetric flasks. After calculating the wts. of samples in solution from volume

and d., each solution was further diluted, with CHCl3, to contain 100-200 γ of combined acetaldehyde/ml. For analysis, 1- and 2-ml. aliquots were hydrolyzed by refluxing with 10% H2SO4 for 15 min., followed by distillation into 5 ml. of freshly prepared 2% NaHSO3. The CHCl3 was separated, then

extracted with 3 10-ml. portions of cold H2O. Exts. were combined with the original aqueous distillate, and diluted to 50 ml. A 1-ml. aliquot was mixed with 8 ml. cold H2SO4-CuSO4 reagent and 0.2 ml. p-phenylphenol reagent (loc. cit.). The tubes were aged 1 hr. in the dark, heated in a 100° water bath for 90 sec., brought to room temperature, then read at 572 m μ . A standard curve was prepared by analyzing known solns. of metaldehyde. One γ of acetaldehyde produced an absorbance of about 0.150.

IT 75-07-0P, Acetaldehyde
RL: PREP (Preparation)

(acetals, determination of, and determination of compds. yielding AcH)

RN 75-07-0 CAPLUS

CN Acetaldehyde (CA INDEX NAME)

 $H_3C-CH=0$

ACCESSION NUMBER: 1952:29719 CAPLUS

DOCUMENT NUMBER: 46:29719
ORIGINAL REFERENCE NO.: 46:5074e-f

TITLE: Treatment of oxidation products
INVENTOR(S): Dice, Henry K.; Mitchell, Robert L.

PATENT ASSIGNEE(S): Celanese Corp. of America

DOCUMENT TYPE: Patent LANGUAGE: Unavailable

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

US 2570217 19511009 US 1949-125834 19491105 <--

AB In the concentration and isolation of products of the partial oxidation of aliphatic hydrocarbons, improved yields of the desired simple compds. are obtained by buffering the aqueous solns. of the crude products to inhibit acetal formation, oxide hydrolysis, Cannizzaro and aldol reactions, and esterification. In each of 3 distns. of the CH2O-rich aqueous solution, a buffer

solution such as aqueous NaOAc is introduced with the reflux to maintain a distillate pH of about 5. The gases flashed off from the aqueous solution during quenching are collected in water and the new solution

fractionated with a similar continuous addition of buffer to maintain a pH of 5-6 during the AcH and acrolein cuts. Cf. preceding abstract

IT · 75-07-0P, Acetaldehyde

RL: PREP (Preparation)

(recovery from hydrocarbon oxidation products)

RN 75-07-0 CAPLUS

CN Acetaldehyde (CA INDEX NAME)

 $H_3C-CH=0$

L16 ANSWER 8 OF 11 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1950:16751 CAPLUS

DOCUMENT NUMBER: 44:16751
ORIGINAL REFERENCE NO.: 44:3307e-g

TITLE: A correlation of plate efficiencies in fractionating

columns

AUTHOR(S): Williams, G. C.; Stigger, E. K.; Nichols, J. H.

CORPORATE SOURCE: Univ. of Louisville, Louisville, KY

SOURCE: Chem. Eng. Progress (1950), 46(No. 1), 7-16

DOCUMENT TYPE: Journal LANGUAGE: Unavailable

AB This report presents data, observations, and conclusions on the over-all efficiencies of a bubble-cap rectification column when operated with 8 different binary distillation systems. Combinations were as follows: (1) EtOH-H2O, (2) MeOH-H2O, (3) benzene-toluene, (4) benzene-CCl4, (5) acetaldehyde-water, (6), acetone-water, (7) MeEtCO-H2O, (8) toluene-pentane. Agreement of the correlation and the data varied with the system and conditions of the test. Low reflux ratios (up to 1.5) produced values in fair agreement, but high reflux ratios and (or) stripping sections varied over a wider range.

IT 75-07-0P, Acetaldehyde
RL: PREP (Preparation)

(distillation apparatus for)

RN 75-07-0 CAPLUS

CN Acetaldehyde (CA INDEX NAME)

L16 ANSWER 9 OF 11 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1948:44440 CAPLUS
DOCUMENT NUMBER: 42.44440

DOCUMENT NUMBER: 42:44440

ORIGINAL REFERENCE NO.: 42:9244i,9245a-b

TITLE: Condensation products of aldehydes with wood-

distillation products containing aryloxy

compounds

INVENTOR(S): Fodor, Miklos

PATENT ASSIGNEE(S): Textilvegyipar Kft.

DOCUMENT TYPE: Patent

LANGUAGE: Unavailable

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

KIND DATE APPLICATION NO. PATENT NO. DATE --------------HU 128789 19411215 HU

AΒ (1) Crude wood tar (100 kg.), 60 kg. 40% formaldehyde, 10 kg. paraformaldehyde, and 4 kg. concentrated HCl are heated until a sample dropped into cold water becomes rigid. The mixture is cooled, and the solid residue separated, treated with steam for several hrs., and dried to give a brown transparent mass, which can be condensed with hexamethylenetetramine or other base that decomps. to formaldehyde only in presence of catalysts. (2) Crude wood tar (100 kg.) is treated with ammonia until it is alkaline to turmeric, 60 kg. 40% formaldehyde and 15 kg. paraformaldehyde are added, and the mixture is boiled under a reflux condenser until a sample dropped into cold water becomes rigid. The mixture is cooled, the solid residue is separated, and uncondensed substances are removed by treatment with benzene or gasoline. Instead of formaldehyde other aldehydes, as acetaldehyde, furfural, instead of paraformaldehyde other polymerized aldehydes, and instead of acid or alkaline catalysts other known catalyst for phenol-aldehyde condensations may be used.

75-07-0P, Acetaldehyde

RL: PREP (Preparation)

(condensation products or polymers of, with wood-distillation products)

RN75-07-0 CAPLUS

CN Acetaldehyde (CA INDEX NAME)

 $H_3C-CH=0$

L16 ANSWER 10 OF 11 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1948:2678 CAPLUS

DOCUMENT NUMBER: 42:2678

ORIGINAL REFERENCE NO.: 42:587i,588a-e TITLE: Acetic anhydride

INVENTOR(S): Hall, Reginald H.; Tuerck, Karl H. W.

PATENT ASSIGNEE(S): Distillers Co., Ltd.

DOCUMENT TYPE: Patent LANGUAGE: Unavailable

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE --------------------GB 589940 19470703 GB 1945-8314 19450404 <--A process for the recovery of Ac20 (I) free of water and substantially AB free of per compds. in yields greater than 90% from the reaction mixture

obtained by the liquid-phase oxidation of AcH (II) with O2, containing in addition to I catalysts, AcOH, acid, water, II, and per compds., by continuously feeding the reaction mixture into the middle of a packed column and effecting fractional distillation therein at pressures of 80-200 mm. and at reflux ratios of between 1/1 and 3/1, at the same time maintaining the liquid at the base of the column above 70°, and withdrawing the concentrated I from the kettle. The volume of the kettle should be sufficient

to

bring about substantially complete (less than 0.1% remaining) decomposition of the per compds. The process prevents loss of II which occurs at low distillation pressures, and requires a smaller condenser surface. Packing holdup should be less than 10%. The effect of various reaction conditions are shown as follows: Run Number, Column pressure mm., Kettle temperature °, Yield of anhydride %, AcOH in kettle liquid %, Time to reduce per compds. to less than 0.1% hrs.; 1, 25, 48, 94, 38, at least 55; 2, 25, 54, 98, 15, 55; 3, 80, 75, 96.3, 6, 0.66; 4, 150, 86, 94.8, 17, 0.079; 5, 150, 79, 87.4, 45.6,; 6, 250, 95, 87.4, 26, 0.016; For example, a reaction product from the oxidation of II, containing I 34.9, AcOH 50.7, H2O 8.0, II 4.3, AcOMe 0.9, a Co and Cu acetate mixture 0.5, and per compds. 0.7% was fractionated at 150-160 mm., in a packed column having 13 theoretical plates and equipped with a kettle. The feed was introduced at the 6th plate from the bottom of the column. The vapor velocity in the column was about 80 cm./sec., the reflux ratio 3:1, and the surface of the packing per theoretical plate 0.12 sq. m. kettle temperature was 83°, the temperature at the feed point 68°, and the temperature at the top of the column 61°. The liquid in the kettle contained I 85.2%, AcOH 14.8, and per compds. 0.0025%. The recovery of I was 94%. The distillate contained AcOH 76.5, II 6.2, and MeOH 2.0%, the residue being H2O. The concentrated anhydrous I may be withdrawn

in the liquid or vapor state and fractionated in a 2nd still. Contains 2 examples.

IT 75-07-0P, Acetaldehyde

RL: PREP (Preparation)

(oxidation of, Ac20 recovery from)

RN 75-07-0 CAPLUS

CN Acetaldehyde (CA INDEX NAME)

 $H_3C-CH=0$

L16 ANSWER 11 OF 11 CAPLUS COPYRIGHT 2007 ACS on STN

ACCESSION NUMBER: 1919:721 CAPLUS

DOCUMENT NUMBER: 13:721
ORIGINAL REFERENCE NO.: 13:134a-d

TITLE: Alcohols, esters

PATENT ASSIGNEE(S): du Pont de Nemours & Co.

DOCUMENT TYPE: Patent LANGUAGE: Unavailable

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE _ _ _ _ ----------GB 119249 19181003 GB 1917-8259 19170609 <--AB Alcs. are prepared by reacting with a halogen hydrocarbon on a salt of a higher fatty acid, e. g., Na stearate, and saponifying the resulting ester. A suitable apparatus is specified. The reaction vessel, e. g., is half filled with stearic acid, the equivalent of solid NaOH is added, the mixture is heated by an oil jacket to 200-40°, and a mixture of pentyl and hexyl chlorides is run in from a suitable vessel. Unchanged chlorides, and olefins produced as by-products, distil over to condenser and

receptacle. The reaction vessel is then cooled to 150°, NaOH is added, and the vessel again heated to saponify the esters. The resulting alcs. are distilled off in steam, and the process is repeated until the stearate in the reaction vessel has become highly charged with salt. Then the mixture is removed, acidified, warmed, and the stearic acid decanted off, and washed for re-use. In a modified construction of apparatus, a reflux condenser is provided to return unchanged chlorides to the reaction vessel, while only the olefins pass off to the condenser.

TT 75-07-0P, Acetaldehyde RL: PREP (Preparation)

(ethyl alc. preparation from)

RN 75-07-0 CAPLUS

CN Acetaldehyde (CA INDEX NAME)

 $H_3C-CH=0$

Refine Search

Search Results -

Terms	Documents
L5 and (562/\$ or 568/\$)	56

US Pre-Grant Publication Full-Text Database US Patents Full-Text Database US OCR Full-Text Database

Database:

EPO Abstracts Database
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IBM Technical Disclosure Bulletins

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L6

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Search History

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DB=F	PGPB,USPT,USOC,EPAB,JPAB,DWPI,TDBD; PLUR=YES; OP=ADJ		
<u>L6</u>	L5 and (562/\$ or 568/\$)	56	<u>L6</u>
. <u>L5</u>	L4 and (density or concentra\$8 or heat\$5 rate or column pressure or feed comp\$9 or reflux or reflux ratio)	208	<u>L5</u>
<u>L4</u>	L3 and distil\$9	223	<u>L4</u>
<u>L3</u>	acetaldehyde same methyl iodide	308	L3
<u>L2</u>	acetaldehyde and methyl iodide	2530	<u>L2</u>
DB=U	JSPT; PLUR=YES; OP=ADJ		
<u>L1</u>	acetaldehyde.ti.	219	<u>L1</u>

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Search Results - Record(s) 1 through 10 of 56 returned.

☐ 1. Document ID: US 20060293537 A1

L6: Entry 1 of 56

File: PGPB

Dec 28, 2006

PGPUB-DOCUMENT-NUMBER: 20060293537

PGPUB-FILING-TYPE:

DOCUMENT-IDENTIFIER: US 20060293537 A1

TITLE: Control method for process of removing permanganate reducing compounds from

methanol carbonylation process

PUBLICATION-DATE: December 28, 2006

INVENTOR-INFORMATION:

NAME

CITY

STATE

COUNTRY

Trueba; David A.

Webster

TX

US

Kulkarni; Shrikant

Houston

TX

US

US-CL-CURRENT: 562/519

Full	Title	Citation	Front	Review	Classification	Date	Reference	Sequences	Attachments	Claims	KWIC	Draw, De
·····												

□ 2. Document ID: US 20060247466 A1

L6: Entry 2 of 56

File: PGPB

Nov 2, 2006

PGPUB-DOCUMENT-NUMBER: 20060247466

PGPUB-FILING-TYPE:

DOCUMENT-IDENTIFIER: US 20060247466 A1

TITLE: Process for the production of acetic acid

PUBLICATION-DATE: November 2, 2006

INVENTOR-INFORMATION:

NAME CITY STATE COUNTRY Zinobile; Raymond J. Houston TX US Scates; Mark O. Houston TXUS Makelki; Jonathan A. Bay City TX US Salado; Manuel Seabrook TX US

US-CL-CURRENT: <u>562/517</u>; <u>568/492</u>

Full Title Citation Front Review Classification Da	te Refere	nce Sequences	Attachments	Claims	KWAC	Draw, De
☐ 3. Document ID: US 20060178528 A1						
L6: Entry 3 of 56	File:	PGPB		Aug	10,	2006

PGPUB-DOCUMENT-NUMBER: 20060178528

PGPUB-FILING-TYPE:

DOCUMENT-IDENTIFIER: US 20060178528 A1

TITLE: Method of controlling acetic acid process

PUBLICATION-DATE: August 10, 2006

INVENTOR-INFORMATION:

NAME CITY STATE COUNTRY Cawood; James M. JR. Houston TXUS Kulkarni; Shrikant U. Houston ΤX US Liu; Lun-Kuang Houston TXUS

US-CL-CURRENT: 562/519

Full Title Citation	Front Review Cla	ssification Date R	ference Sequences	: Attachments	Claims K	MC Draw De
					· · · · · · · · · · · · · · · · · · ·	
☐ 4. Docume	nt ID: US 20060	011462 A1				
L6: Entry 4 of 5	66	Fi]	e: PGPB		Jan 19	2006

PGPUB-DOCUMENT-NUMBER: 20060011462

PGPUB-FILING-TYPE:

DOCUMENT-IDENTIFIER: US 20060011462 A1

TITLE: Process for producing carboxylic acid and system for producing the same

PUBLICATION-DATE: January 19, 2006

INVENTOR-INFORMATION:

NAME CITY STATE COUNTRY
Horiguchi; Akira Osaka JP
Ina; Tomohide Hyogo JP
Miura; Hiroyuki Hyogo JP

US-CL-CURRENT: 203/29; 202/153, 202/154, 202/155, 203/60, 203/74, 562/6

tachments Claims KMC Draw De

☐ 5. Document ID: US 20050209483 A1

L6: Entry 5 of 56

File: PGPB

Sep 22, 2005

PGPUB-DOCUMENT-NUMBER: 20050209483

PGPUB-FILING-TYPE: new

DOCUMENT-IDENTIFIER: US 20050209483 A1

TITLE: Utilization of acetic acid reaction heat in other process plants

PUBLICATION-DATE: September 22, 2005

INVENTOR-INFORMATION:

NAME CITY STATE COUNTRY

Bhaskaran, Subramanian Singapore SG Sekar, Angadu Krishnamoorthy A.K Singapore SG

US-CL-CURRENT: <u>560/241</u>; <u>562/519</u>

Full	Title Citation	Front	Review	Classification	Date	Reference	Sequences	Attachments	Claims	KWIC	Draw, De
		· · · · · · · · · · · · · · · · · · ·				•					
	6. Docume	ent ID:	US 20	050197513	A 1						
L6: E	ntry 6 of	56				File: 1	PGPB		Sep	8,	2005

PGPUB-DOCUMENT-NUMBER: 20050197513

PGPUB-FILING-TYPE: new

DOCUMENT-IDENTIFIER: US 20050197513 A1

TITLE: CONTROL METHOD FOR PROCESS OF REMOVING PERMANGANATE REDUCING COMPOUNDS FROM METHANOL CARBONYLATION PROCESS

PUBLICATION-DATE: September 8, 2005

INVENTOR-INFORMATION:

NAME CITY STATE COUNTRY

Trueba, David A. Webster TX US
Kulkarni, Shrikant Houston TX US

US-CL-CURRENT: 568/492

Full	Title Citation	Front Re	view Clas	sification	Date	Referenc	Sequences	Attachments	Claims	KWIC	Draw, De
	7. Docume	nt ID: US	S 20050	197509	A1						
L6: En	try 7 of 5	66				File:	PGPB		Sep	8.	2005

PGPUB-DOCUMENT-NUMBER: 20050197509

PGPUB-FILING-TYPE: new

DOCUMENT-IDENTIFIER: US 20050197509 A1

TITLE: REMOVAL OF PERMANGANATE REDUCING COMPOUNDS FROM METHANOL CARBONYLATION PROCESS STREAM

Sep 8, 2005

PUBLICATION-DATE: September 8, 2005

INVENTOR-INFORMATION:

NAME CITY STATE COUNTRY Picard, Wayne D. Houston TXUS Talancon, Jose J. Arturo Col. Heroes de Churubusco TX MX Trueba, David A. Webster ТX US Scates, Mark O. Houston ТX US Zinobile, Raymond J. Houston US

US-CL-CURRENT: <u>562/608</u>

Full Title Citation Front R	eview Classification Date	Referenc	e Sequences	Attachments	Claims	KWIC	Draw, De
□ 8. Document ID: U	JS 20050197508 A1						
L6: Entry 8 of 56		File:	PGPB		Sep	8,	2005

PGPUB-DOCUMENT-NUMBER: 20050197508

PGPUB-FILING-TYPE: new

DOCUMENT-IDENTIFIER: US 20050197508 A1

TITLE: REMOVAL OF PERMANGANATE REDUCING COMPOUNDS FROM METHANOL CARBONYLATION

PROCESS STREAM

PUBLICATION-DATE: September 8, 2005

INVENTOR-INFORMATION:

NAME CITY STATE COUNTRY Scates, Mark O. Houston ΤX US Trueba, David A. Webster TX US Zinobile, Raymond J. Houston ΤX US

US-CL-CURRENT: <u>562/608</u>

Full	Title Citation	Front	Review	Classification	Date	Reference	Sequences	Attachments	Claims	KWIC	Draw, De
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	9. Docume	nt ID:	US 20	050197506	A1						
L6: E	Entry 9 of 5	56				File: F	GPB		Sep	8,	2005

PGPUB-DOCUMENT-NUMBER: 20050197506

PGPUB-FILING-TYPE: new

DOCUMENT-IDENTIFIER: US 20050197506 A1

TITLE: Process for Producing Acetic Acid

PUBLICATION-DATE: September 8, 2005

INVENTOR-INFORMATION:

NAME

CITY

STATE

COUNTRY

Scates, Mark O.

Houston

TX

US

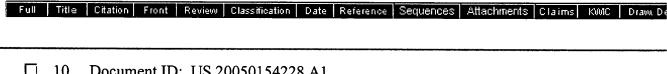
Trueba, David A.

Webster

TX

US

US-CL-CURRENT: <u>562/519</u>; <u>562/608</u>



☐ 10. Document ID: US 20050154228 A1

L6: Entry 10 of 56

File: PGPB

Jul 14, 2005

PGPUB-DOCUMENT-NUMBER: 20050154228

PGPUB-FILING-TYPE: new

DOCUMENT-IDENTIFIER: US 20050154228 A1

TITLE: Method and apparatus for controlling feed of gaseous reaction component

PUBLICATION-DATE: July 14, 2005

INVENTOR-INFORMATION:

NAME

CITY

STATE

COUNTRY

Nakajima, Hidehiko

Himeji-shi, Hyogo

JΡ

Harano, Yoshiyuki

Ako-gun Hyogo

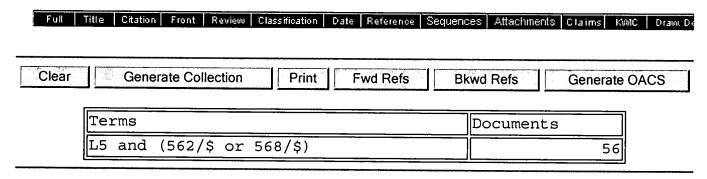
JΡ

Yamada, Manabu

Himeji-shi Hyogo

JP

US-CL-CURRENT: <u>562/519</u>; <u>422/187</u>



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Search Results - Record(s) 11 through 20 of 56 returned.

☐ 11. Document ID: US 20040122257 A1

L6: Entry 11 of 56

File: PGPB

Jun 24, 2004

PGPUB-DOCUMENT-NUMBER: 20040122257

PGPUB-FILING-TYPE: new

DOCUMENT-IDENTIFIER: US 20040122257 A1

TITLE: Low water methanol carbonylation process for high acetic acid production and

for water balance control

PUBLICATION-DATE: June 24, 2004

INVENTOR-INFORMATION:

NAME CITY STATE COUNTRY Cheung, Hosea Corpus Christi TX US Huckman, Michael E. Corpus Christi ΤX US Torrence, G. Paull League City TX US

US-CL-CURRENT: 562/519

Full	Title	Citation	Front	Review	Classification	Date	Reference	≘ Sequences	Attachments	Claims	KWIC	Drawu	D٠
													_
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	12.	Docum	nent ID): US 2	003019971	1 A1							
L6:	Entry	12 of	56				File:	PGPB		Oct	23,	2003	

PGPUB-DOCUMENT-NUMBER: 20030199711

PGPUB-FILING-TYPE: new

DOCUMENT-IDENTIFIER: US 20030199711 A1

TITLE: Oxidation treatment of a recycle stream in production of acetic acid by

methanol carbonylation

PUBLICATION-DATE: October 23, 2003

INVENTOR-INFORMATION:

NAME	CITY	STATE	COUNTRY
Broussard, Jerry A.	Marietta	GA	US
Cheung, Hung-Cheun	Corpus Christi	TX	US
Houliston, Stephen Andrew	Ottumwa	IA	US
Huckman, Michael E.	Corpus Christi	TX	US

McKarns MacAtangay, Peggy	Corpus Christi	TX	US
Singh, Madan	Corpus Christi	TX	US
Karnilaw, Michael L.	Houston	TX	US
Torrence, G.Paull	Corpus Christi	TX	US

US-CL-CURRENT: <u>562/519</u>

Full	Title	Citation	Front	Review	Classification	Date	Reference	Sequences	Attachments	Claims	KWIC	Draw, De
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!	13.	Docume	ent ID	: US 2	003013023	8 AI						
L6: E	ntry	13 of	56				File: Po	GPB	•	Jul	10,	2003

PGPUB-DOCUMENT-NUMBER: 20030130238

PGPUB-FILING-TYPE: new

DOCUMENT-IDENTIFIER: US 20030130238 A1

TITLE: Allenic aryl sulfonamide hydroxamic acids as matrix metalloproteinase and

tace inhibitors

PUBLICATION-DATE: July 10, 2003

INVENTOR-INFORMATION:

NAME CITY STATE COUNTRY
Sandanayaka, Vincent Premaratna Northboro MA US
Delos Santos, Efren Guillermo Nanuet NY US

US-CL-CURRENT: 514/114; 514/217.12, 514/357, 514/408, 514/575, 540/609, 546/336, 548/566, 558/166, 562/623

Full Title	Citation Front	Review	Classification	Date	Reference	Sequences	Attachments	Claims	KWIC	Drawu De
☐ 14. L6: Entry	Document II	D: US 2	2002015174		File: P			· · · · · · · · · · · · · · · · · · ·		

PGPUB-DOCUMENT-NUMBER: 20020151746

PGPUB-FILING-TYPE: new

DOCUMENT-IDENTIFIER: US 20020151746 A1

TITLE: Low energy carbonylation process

PUBLICATION-DATE: October 17, 2002

INVENTOR-INFORMATION:

NAME	CITY	STATE	COUNTRY
Scates, Mark O.	Friendswood	TX	US
Blay, George A.	Corpus Christi	TX	US
Torrence, G. Paull	Corpus Christi	TX	US
Broussard, Jerry A.	Corpus Christi	TX	US

US-CL-CURRENT: 562/519

Full Title Citation Front Review Classification Date Reference Sequences Attachments Claims KMC Draw. De

☐ 15. Document ID: US 7196227 B2

L6: Entry 15 of 56

File: USPT

Mar 27, 2007

US-PAT-NO: 7196227

DOCUMENT-IDENTIFIER: US 7196227 B2

TITLE: Process for the recovery of phosphorus and iodine containing catalyst

components

PRIOR-PUBLICATION:

DOC-ID

DATE

US 20040163840 A1

August 26, 2004

Full Title Citation Front Review Classification Date Reference **Sequences Attachments** Claims KMC Draw. De

☐ 16. Document ID: US 7005541 B2

L6: Entry 16 of 56

File: USPT

Feb 28, 2006

US-PAT-NO: 7005541

DOCUMENT-IDENTIFIER: US 7005541 B2

TITLE: Low water methanol carbonylation process for high acetic acid production and

for water balance control

PRIOR-PUBLICATION:

DOC-ID

DATE

US 20040122257 A1

June 24, 2004

Full Title Citation Front Review Classification Date Reference Sequences Attachments Claims KMC Draw. De

☐ 17. Document ID: US 6667418 B2

L6: Entry 17 of 56

File: USPT

Dec 23, 2003

US-PAT-NO: 6667418

DOCUMENT-IDENTIFIER: US 6667418 B2

TITLE: Oxidation treatment of a recycle stream in production of acetic acid by

methanol carbonylation

Full Title Citation Front Review Classification Date Reference Sequences Attachments Claims KMC Draw. De

☐ 18. Document ID: US 6657078 B2 L6: Entry 18 of 56 File: USPT Dec 2, 2003 US-PAT-NO: 6657078 DOCUMENT-IDENTIFIER: US 6657078 B2 TITLE: Low energy carbonylation process Full Title Citation Front Review Classification Date Reference Citation Claims KMC Draw De ☐ 19. Document ID: US 6573403 B1 L6: Entry 19 of 56 File: USPT Jun 3, 2003 US-PAT-NO: 6573403 DOCUMENT-IDENTIFIER: US 6573403 B1 TITLE: Process for production of acetic acid Full Title Citation Front Review Classification Date Reference Control Review Claims KMC Draw Do ☐ 20. Document ID: US 6552221 B1 L6: Entry 20 of 56 File: USPT Apr 22, 2003 US-PAT-NO: 6552221 DOCUMENT-IDENTIFIER: US 6552221 B1 ** See image for Certificate of Correction ** TITLE: Process control for acetic acid manufacture Full Title Citation Front Review Classification Date Reference Control Attackments Claims KMC Draw Do Clear Generate Collection Print Fwd Refs **Bkwd Refs Generate OACS** Terms Documents L5 and (562/\$ or 568/\$) 56 **Display Format:** -Change Format Previous Page Next Page Go to Doc#

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Search Results - Record(s) 21 through 30 of 56 returned.

☐ 21. Document ID: US 6346645 B1

L6: Entry 21 of 56

File: USPT

Feb 12, 2002

US-PAT-NO: 6346645

DOCUMENT-IDENTIFIER: US 6346645 B1

TITLE: Adsorptive removal of carbonyl impurities from oxygenated organic liquids

Full Title Citation Front Review Classification Date Reference Structures Attachments Claims KMC Draw. De

☐ 22. Document ID: US 6339171 B1

L6: Entry 22 of 56

File: USPT

Jan 15, 2002

US-PAT-NO: 6339171

DOCUMENT-IDENTIFIER: US 6339171 B1

TITLE: Removal or reduction of permanganate reducing compounds and alkyl iodides

from a carbonylation process stream

Full Title Citation Front Review Classification Date Reference Carbon Company Claims KWC Draw. Do

☐ 23. Document ID: US 6323364 B1

L6: Entry 23 of 56

File: USPT

Nov 27, 2001

US-PAT-NO: 6323364

DOCUMENT-IDENTIFIER: US 6323364 B1

TITLE: Rhodium/inorganic iodide catalyst system for methanol carbonylation process

with improved impurity profile

Full Title Citation Front Review Classification Date Reference Reference Reference Claims KWIC Draw, De

☐ 24. Document ID: US 6251256 B1

L6: Entry 24 of 56

File: USPT

Jun 26, 2001

US-PAT-NO: 6251256

DOCUMENT-IDENTIFIER: US 6251256 B1

TITLE: Process for electrochemical oxidation of an aldehyde to an ester



☐ 25. Document ID: US 6211405 B1

L6: Entry 25 of 56

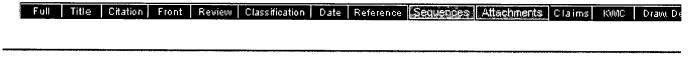
File: USPT

Apr 3, 2001

US-PAT-NO: 6211405

DOCUMENT-IDENTIFIER: US 6211405 B1

TITLE: Addition of iridium to the rhodium/inorganic iodide catalyst system



☐ 26. Document ID: US 6143930 A

L6: Entry 26 of 56

File: USPT

Nov 7, 2000

US-PAT-NO: 6143930

DOCUMENT-IDENTIFIER: US 6143930 A

TITLE: Removal of permanganate reducing compounds and alkyl iodides from a

carbonylation process stream

Full	Title	Citation	Front	Review	Classification	Date	Reference	Sequences	Attachments	Claims	KWIC	Draw. De
			•									

☐ 27. Document ID: US 6066762 A

L6: Entry 27 of 56

File: USPT

May 23, 2000

Apr 18, 2000

US-PAT-NO: 6066762

DOCUMENT-IDENTIFIER: US 6066762 A

TITLE: Process for the production of carbonyl compound

1	Full	Title	Citation Front	Review	Classification	Date	Reference	Sequences	Attachments	Claims	KWIC	Draw, De

		28.	Document II	D: US 6	051590 A					•		

File: USPT

US-PAT-NO: 6051590

DOCUMENT-IDENTIFIER: US 6051590 A

L6: Entry 28 of 56

TITLE: Immunosuppressant tricyclic compounds



☐ 29. Document ID: US 6017455 A

L6: Entry 29 of 56

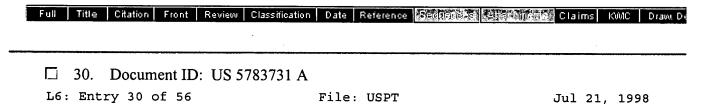
File: USPT

Jan 25, 2000

US-PAT-NO: 6017455

DOCUMENT-IDENTIFIER: US 6017455 A

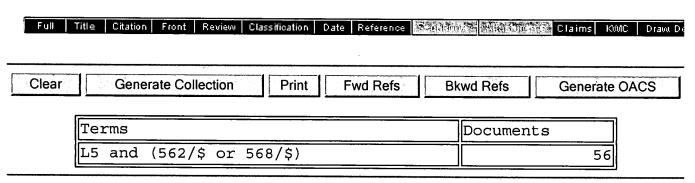
TITLE: Porous membrane



US-PAT-NO: 5783731

DOCUMENT-IDENTIFIER: US 5783731 A

TITLE: Removal of carbonyl impurities from a carbonylation process stream



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First Hit Clear Generate Collection Print Fwd Refs Bkwd Refs

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Search Results - Record(s) 31 through 40 of 56 returned.

☐ 31. Document ID: US 5756836 A

L6: Entry 31 of 56

File: USPT

May 26, 1998

US-PAT-NO: 5756836

DOCUMENT-IDENTIFIER: US 5756836 A

** See image for Certificate of Correction **

TITLE: Process for producing highly purified acetic acid

Full Title Citation Front Review Classification Date Reference Section Attachine Claims KWIC Draw. De

☐ 32. Document ID: US 5723660 A

L6: Entry 32 of 56

File: USPT

Mar 3, 1998

US-PAT-NO: 5723660

DOCUMENT-IDENTIFIER: US 5723660 A

** See image for Certificate of Correction **

TITLE: Process for producing acetic acid

Full Title Citation Front Review Classification Date Reference Alle Control Claims KWIC Draw. Do

☐ 33. Document ID: US 5648531 A

L6: Entry 33 of 56

File: USPT

Jul 15, 1997

US-PAT-NO: 5648531

DOCUMENT-IDENTIFIER: US 5648531 A

TITLE: Process for producing acetic anhydride alone or acetic anhydride and acetic

acid

Full Title Citation Front Review Classification Date Reference Security Attackments Claims KMC Draw De

☐ 34. Document ID: US 5625095 A

L6: Entry 34 of 56

File: USPT

Apr 29, 1997

US-PAT-NO: 5625095

DOCUMENT-IDENTIFIER: US 5625095 A

TITLE: Process for producing high purity acetic acid

Full Title Citation Front Review Classification Date Reference Sequences Attachments Claims KMC Draw Do

☐ 35. Document ID: US 5599976 A

L6: Entry 35 of 56

File: USPT

Feb 4, 1997

US-PAT-NO: 5599976

DOCUMENT-IDENTIFIER: US 5599976 A

TITLE: Recovery of acetic acid from dilute aqueous streams formed during a carbonylation process

Full Title Citation Front Review Classification Date Reference Sequences Attachments Claims KMC Draw. De

☐ 36. Document ID: US 5488143 A

L6: Entry 36 of 56

File: USPT

Jan 30, 1996

US-PAT-NO: 5488143

DOCUMENT-IDENTIFIER: US 5488143 A

TITLE: Processes for the carbonylation of methanol to form acetic acid, methyl acetate and acetic anhydride

Full Title Citation Front Review Classification Date Reference Sequences Altachments Claims KWIC Draw. Do

☐ 37. Document ID: US 5371286 A

L6: Entry 37 of 56

File: USPT

Dec 6, 1994

US-PAT-NO: 5371286

DOCUMENT-IDENTIFIER: US 5371286 A

TITLE: Removal of carbonyl impurities from a carbonylation process stream

Full Title Citation Front Review Classification Date Reference Sequences Attachments Claims KWC Draw. De

☐ 38. Document ID: US 5362365 A

L6: Entry 38 of 56

File: USPT

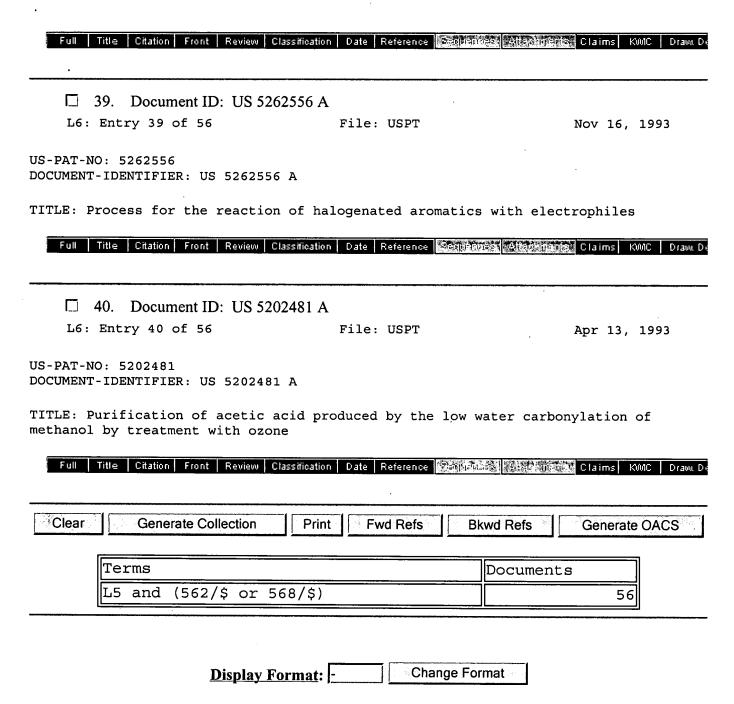
Nov 8, 1994

US-PAT-NO: 5362365

DOCUMENT-IDENTIFIER: US 5362365 A

TITLE: Purification of acetic anhydride or acetic anhydride and acetic acid using

ozone



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Search Results - Record(s) 41 through 50 of 56 returned.

☐ 41. Document ID: US 5155266 A

L6: Entry 41 of 56

File: USPT

Oct 13, 1992

US-PAT-NO: 5155266

DOCUMENT-IDENTIFIER: US 5155266 A

TITLE: Purification of acetic acid with ozone in the presence of an oxidation

catalyst

Full Title Citation Front Review Classification Date Reference Sequences Attachnents Claims KWIC Draw. De

☐ 42. Document ID: US 5155265 A

L6: Entry 42 of 56

File: USPT

Oct 13, 1992

US-PAT-NO: 5155265

DOCUMENT-IDENTIFIER: US 5155265 A

TITLE: Purification of acetic acid with ozone followed by treatment with activated

carbon and/or an ion-exchange resin

Full Title Citation Front Review Classification Date Reference Contact Attachments Claims KMC Draw. De

☐ 43. Document ID: US 5138093 A

L6: Entry 43 of 56

File: USPT

Aug 11, 1992

US-PAT-NO: 5138093

DOCUMENT-IDENTIFIER: US 5138093 A

TITLE: Process for preparing ethylidene diacetate

Full Title Citation Front Review Classification Date Reference **Company Made Note:** Claims KMC Draw. Do

☐ 44. Document ID: US 4978778 A

L6: Entry 44 of 56

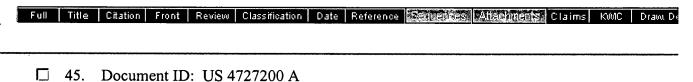
File: USPT

Dec 18, 1990

US-PAT-NO: 4978778

DOCUMENT-IDENTIFIER: US 4978778 A

TITLE: Process for producing vinyl acetate



L6: Entry 45 of 56

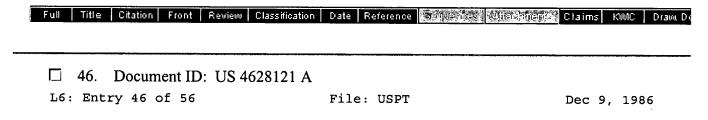
File: USPT

Feb 23, 1988

US-PAT-NO: 4727200

DOCUMENT-IDENTIFIER: US 4727200 A

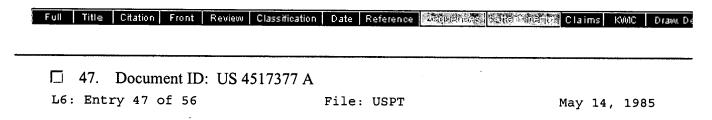
TITLE: Alcohol homologation



US-PAT-NO: 4628121

DOCUMENT-IDENTIFIER: US 4628121 A

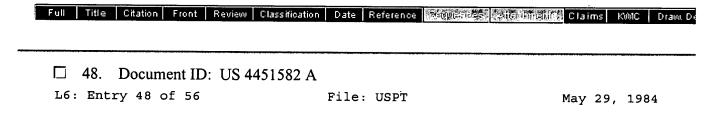
TITLE: Preparation of acetaldehyde



US-PAT-NO: 4517377

DOCUMENT-IDENTIFIER: US 4517377 A

TITLE: Process for producing vinyl acetate



US-PAT-NO: 4451582

DOCUMENT-IDENTIFIER: US 4451582 A

TITLE: Preparation of insoluble, only slightly swellable polymers of basic vinylheterocyclic compounds



☐ 49. Document ID: US 4399001 A

L6: Entry 49 of 56

File: USPT

Aug 16, 1983

US-PAT-NO: 4399001

DOCUMENT-IDENTIFIER: US 4399001 A

TITLE: Separating off organic iodine compounds from acetaldehyde-free carbonylation

products of methanol, methyl acetate and dimethyl ether



☐ 50. Document ID: US 4383894 A

L6: Entry 50 of 56

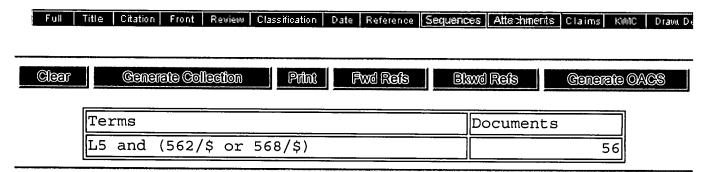
File: USPT

May 17, 1983

US-PAT-NO: 4383894

DOCUMENT-IDENTIFIER: US 4383894 A

TITLE: Removal of organic iodine compounds from acetaldehyde



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Search Results - Record(s) 51 through 56 of 56 returned.

☐ 51. Document ID: US 4302611 A

L6: Entry 51 of 56

File: USPT

Nov 24, 1981

US-PAT-NO: 4302611

DOCUMENT-IDENTIFIER: US 4302611 A

TITLE: Preparation of acetaldehyde

Full Title Citation Front Review Classification Date Reference Sequences Attachments Claims KMC Draw. De

☐ 52. Document ID: US 4269788 A

L6: Entry 52 of 56

File: USPT

May 26, 1981

US-PAT-NO: 4269788

DOCUMENT-IDENTIFIER: US 4269788 A

TITLE: Phenyl-cyclohexadiene-alkylamine derivatives

Full Title Citation Front Review Classification Date Reference Sequences Attachments Claims KWIC Draw. De

☐ 53. Document ID: US 4252748 A

L6: Entry 53 of 56

File: USPT

Feb 24, 1981

US-PAT-NO: 4252748

DOCUMENT-IDENTIFIER: US 4252748 A

TITLE: Recovery of acetone produced by carbonylation

Full Title Citation Front Review Classification Date Reference **Sequences Attachme**nts Claims KMC Draw. De

☐ 54. Document ID: US 4252741 A

L6: Entry 54 of 56

File: USPT

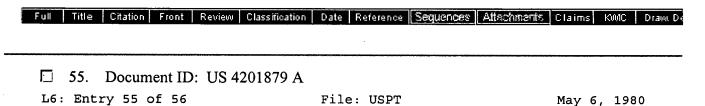
Feb 24, 1981

US-PAT-NO: 4252741

DOCUMENT-IDENTIFIER: US 4252741 A

** See image for Certificate of Correction **

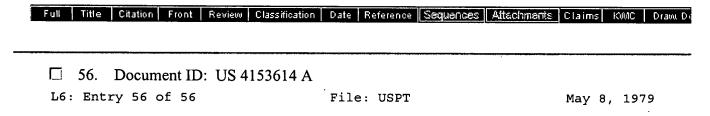
TITLE: Carbonylation with Group VIII noble metal catalysts



US-PAT-NO: 4201879

DOCUMENT-IDENTIFIER: US 4201879 A

TITLE: Hydroquinones

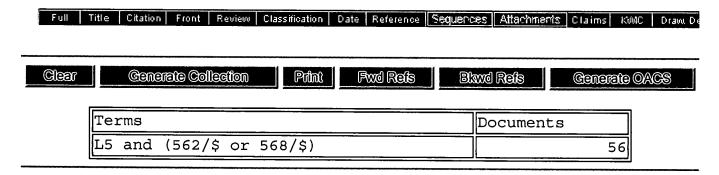


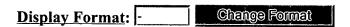
US-PAT-NO: 4153614

DOCUMENT-IDENTIFIER: US 4153614 A

TITLE: Synthesis of (S)-(+)-6-hydroxy-2,5,7,8-tetramethylchroman-2-methanol and

intermediates therein





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